Category 1 (respiratory organs, cardiovascular)

Safety Data Sheet

1. Chemical product and company identification

Product name : HYDROFLUORIC ACID (50%)

Company information

Name of manufacturer : KANTO CHEMICAL CO., INC.

Address : 2-1, Nihonbashi, Muromachi 2-Chome, Chuo-Ku, Tokyo, 103-0022, Japan

Name of section : Electronic materials division technical department

Telephone number : +81-3-6214-1080 Facsimile number : +81-3-3241-1043 Mail address : el-info@kanto.co.jp

Reference No : GE00172 1.2

Recommended uses and : Electronic chemicals

restrictions

2. Hazards identification

GHS classification

Physical hazards Corrosive to metals Category 1 Health hazards Acute toxicity (oral) Category 2 Acute toxicity (dermal) Category 3 Acute toxicity Category 1

(inhalation:dust/mist)

Skin corrosion/irritation Category 1B Serious eye damage/eye Category 1

irritation

Specific target organ toxicity

(single exposure)

Specific target organ toxicity

(repeated exposure)

Category 1 (teeth, bone)

Environmental

hazards

Aquatic acute Category 3

Hazard pictograms







Signal word Danger

Hazard statements May be corrosive to metals

Fatal if swallowed or if inhaled

Toxic in contact with skin

Causes severe skin burns and eye damage

Causes damage to organs (respiratory organs, cardiovascular) Causes damage to organs (teeth, bone) through prolonged or

repeated exposure Harmful to aquatic life

Precautionary statements

Page

Prevention : Keep only in original container.

Do not breathe mist/vapors.

Wash hands, forearms and face thoroughly after handling. Do not eat, drink or smoke when using this product. Use only outdoors or in a well-ventilated area.

Avoid release to the environment.

Wear protective gloves/protective clothing/eye protection/face protection.

[In case of inadequate ventilation] wear respiratory protection.

IF SWALLOWED: Immediately call a POISON CENTER or doctor.

IF SWALLOWED: Rinse mouth. Do not induce vomiting.

IF ON SKIN: Wash with plenty of water.

IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water.

IF INHALED: Remove person to fresh air and keep comfortable for breathing.

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

IF exposed or concerned: Call a POISON CENTER or doctor.

Immediately call a POISON CENTER or doctor.

Call a POISON CENTER or doctor if you feel unwell. Get medical advice/attention if you feel unwell.

Take off immediately all contaminated clothing and wash it before

Absorb spillage to prevent material-damage.

Store in a well-ventilated place. Keep container tightly closed.

Store locked up.

Disposal : Dispose of contents/container to hazardous or special waste

collection point, in accordance with local, regional, national

and/or international regulation.

3. Composition/information on ingredients

Distinction of substance or : Substance

mixture

Storage

Response

Chemical name	Concentration (%)	Formula	TSCA	EC-No.	CAS RN
Hydrofluoric acid	50 (as F 48)	HF	Listed	231-634-8	7664-39-3

4. First aid measures

First aid measures

First-aid measures after inhalation

First-aid measures after skin contact

: Remove victim to fresh air and keep at rest in a position comfortable for breathing. Immediately get medical treatment.

Remove contaminated clothing and shoes immediately. Flush affected areas with plenty of water immediately. If possible, apply calcium gluconate jelly on affected areas. Get medical attention.

First-aid measures after eye contact

First-aid measures after ingestion

: Wash the affected areas under running water for at least 15 minutes. Get medical treatment.

: Give the victim milk or 5% calcium gluconate water solution. Get medical attention.



Personal Protection in First Aid and Measures : Rescuers should wear proper protective equipment like rubber gloves, goggles.

Most Important Symptoms/Effects

Symptoms/effects

: The substance is very corrosive to the eyes, skin and respiratory tract. Inhalation of the vapour may cause lung oedema.

5. Fire fighting measures

Suitable extinguishing media

: This product is noncombustible.

Unsuitable extinguishing media

: None

Firefighting instructions

: Move containers from fire area if it can be done without risk, if not possible, apply water from a safe distance to cool and

protect surrounding area.

Personal protection (Emergency

response)

: Firefighters should wear protective equipment.

6. Accidental release measures

Personal Precautions, Protective Equipment and Emergency Procedures

General measures

: Wear proper protective equipment and avoid contact with skin and inhalation of vapor. Conduct operations from upwind and evacuate people downwind. Keep away personnel except for authorized ones from spillage area by stretching ropes.

Environmental precautions

Environmental precautions

: Attention should be given to avoid discharge of spilled product into rivers and resulting environmental damage. When diluting spill with large amounts of water, discharge of untreated wastewater into the environment must be avoided.

Methods and Equipment for Containment and Cleaning up

For containment

: Collect the spillage as much as possible to switable empty container. Or else dilute with water gradually and neutralize with calcium hydroxide solution or sodium carbonate solution then wash thoroughly with water.

7. Handling and storage

Handling

Technical measures

: Wear proper protective equipment to avoid contact with skin or inhalation of vapor.

Precautions for safe handling

Use with an enclosed system or a local exhaust ventilation. Use in well-ventilated areas.

Because this chemical corrodes many kinds of metal, glass and concretes, choose appropriate materials.

The substance is acidic. Avoid contact with alkaline substances.

Storage

Storage conditions
Material used in
packaging/containers

: Store in a dark, cool place and tightly closed.

: Polyethylene, fluorine resin.

Do not use many kinds of metals like carbon steel, low-alloy

steel, nickel, glass.



8. Exposure controls / Personal protection equipment

ACGIH TWA	0.5 ppm (as F)
ACGIH Ceiling	2 ppm (as F)
Remark (ACGIH)	Skin

Appropriate engineering

controls

: Use with an enclosed system or a local exhaust ventilation.

Protective equipment

Respiratory protection : Chemical cartridge respirator with acids vapor cartage or airline

respirator

Hand protection : Impervious protective gloves

Eye protection : Safety goggles

Skin and body protection : Protective clothing, protective boots

9. Physical and chemical properties

Physical state : Liquid
Color : Colorless.
Odor : Pungent.

pH : Strong acidity

Melting point : -35 ° C

Freezing point : No data available

Boiling point : 104 $^{\circ}$ C

Flash point Non flammable. Auto-ignition temperature : Non flammable. Decomposition temperature No data available Flammability : Non flammable. Vapor pressure : No data available Relative density : No data available Density 1.18 g/cm³ (20°C) No data available Relative gas density Solubility : Water: Miscible. Partition coefficient n-: No data available

octanol/water (log Pow)

Explosive limits (vol %) : No data available
Viscosity, kinematic: : No data available
Particle characteristics : No data available

10. Stability and reactivity

Reactivity : This substance is a strong acid and reacts with alkaline substances

and many metals. Glass also corrodes.

Chemical stability : Stable under normal conditions. Fuming in air.

Possibility of hazardous : Corrodes most metals except gold and platinum. At that time,

reactions flammable hydrogen gas may be generated.

Conditions to avoid : Light, heat.

Incompatible materials : Alkaline substances, metals. Hazardous decomposition : Hydrogen fluoride, fluorine.

products

11. Toxicological information

Acute toxicity (oral) Fatal if swallowed

> It has been reported that when hydrogen fluoride is taken orally, the lethal dose in humans is 1.5 g or 20 mg / kg. Based on the

above, it was classified into category 2.

Acute toxicity (dermal) Toxic in contact with skin

> Chemical burns with a body surface area of 7% caused by 50-70% hydrofluoric acid have been reported to be fatal. Based on the

above, it was classified into category 3.

Acute toxicity (inhalation) : No classification (gas)

Classification not possible (vapor)

Fatal if inhaled

Acute toxicity (mist) -

Description

: It has been reported that the lethal concentration of hydrogen fluoride fume is 50-250 ppm (mist, 4-hour equivalent: 0.0008-0.004 mg / L) or higher. Based on the above, it was classified into

category 1.

Skin corrosion/irritation Causes severe skin burns

> When hydrofluoric acid adheres to the skin and mucous membranes, it causes strong painful corrosion locally, causing systemic effects and death in some cases. It is said that if hydrofluoric acid adheres to 50-100 cm2 of the skin, it is necessary to be hospitalized, and if it is 100 cm2 or more, it should be treated with ICU. In addition, in a skin irritation test in rabbits (OECD TG404), significant crusting with destruction of skin tissue occurred and did not recover within 14 days. Based on the above, it

was classified into category 1B.

Serious eye damage/irritation Causes serious eye damage

> The skin corrosion / irritatation section of this substance is category 1B. Based on the above, it was classified into category 1.

Respiratory sensitization

Skin sensitization

Germ cell mutagenicity

Classification not possible Classification not possible

: No classification

It could not be concluded from the results of in vivo tests with hydrofluoric acid (a mouse dominant lethal test, a mutual translocation tests with mouse testes, a chromosomal aberration test with rat bone marrow cells). It was described that evaluation was possible by tests with sodium fluoride, which showed no chromosomal aberration. However, although abnormality was observed in anaphase cells, since fluoride ions do not bind to DNA covalently, and do not form DNA adducts, it is thought to be a secondary effect. Therefore, it is concluded that inorganic fluoride does not induce chromosomal damage in vivo. As for in

vitro, a bacterial reverse mutation test was negative. Carcinogenicity Classification not possible Reproductive toxicity Classification not possible

STOT-single exposure Causes damage to organs (respiratory organs, cardiovascular)

As a result of percutaneous exposure to 80% hydrofluoric acid, cases of death due to "dyspnea with chest pain" and "severe hypocalcemia" have been reported. Thus, it was classified into

category 1 (respiratory organs, cardiovascular).

STOT-repeated exposure : Causes damage to organs (teeth, bone) through prolonged or repeated

exposure

In humans, it has been reported that fluoride accumulates in bone and tooth buds, and chronic exposure to high levels of fluoride and hydrogen fluoride causes skeletal fluorosis. Based on the above, it

was classified into category 1 (teeth, bone).

Aspiration hazard : Classification not possible

12. Ecological information

Ecotoxicity

Aquatic acute : Harmful to aquatic life

Chaetogammarus marinus EC50=38.28mg F/L/96h (as sodium fluoride)

Aquatic chronic : No classification

Oryzias latipes NOEC≥9.9mg F/L/28-day (as sodium fluoride)

Persistence and degradability

No additional information available

Bioaccumulative potential

No additional information available

Mobility in soil

No additional information available

Hazardous to the ozone layer

Ozone : Classification not possible

13. Disposal considerations

Ecological waste information : Add

Add the chemical in a large amount of calcium hydroxide solution gradually to precipitate calcium fluoride. Filter the precipitation and bury in a landfill site approved for hazardous waste disposal. Or entrust approved waste disposal companies with the disposal.

 $\langle Note \rangle$

*The pH of the neutralization should be above 8.5. The precipitation does not form completely below pH 8.5.

*If added rapidly calcium hydroxide solution in the chemical, generate heat and the chemical may fly in all directions.

*As unreacted toxic gas may be emitted at the disposal work, wear appropriate protective equipment. Even a small amount of

the gas has the toxicity.

Contaminated container and

packaging

: In case of disposal of empty bottles, dispose bottles after removing the content thoroughly.

14. Transport information

International Regulations Transport by sea(IMDG)

UN-No. (IMDG)

: 1790

Proper Shipping Name (IMDG) : HYDROFLUORIC ACID

Packing group (IMDG) : II

Revision date: 3/7/2024

: 8 (6.1) Transport hazard class(es)

(IMDG)

Air transport(IATA)

UN-No. (IATA) 1790

Proper Shipping Name (IATA) Hydrofluoric acid

Packing group (IATA) IITransport hazard class(es) 8 (6.1)

(IATA)

Marine pollutant : Not applicable

MFAG-No 157

15. Regulatory information

Regulatory information with regard to this substance in your country or region should be examined by your own responsibility.

16. Other information

Data sources : Handbook of dangeroous and hazardous chemicals, Japan

Industrial Safety & Health Association. (2000-2001) . Dangerous Properties of Industrial Materials, 6th ed.

N. I. Sax Van Nostrand Reinhold Company (1984) .

Handbook of Dangerous Substances Springer-Verlag Tokyo

(1991).

Handbook of 17322 Chemical Products, The Chemical Daily Co.

(2022).

Handbook of Poisonous and Deleterious substances, revised and

enlarged edition, Yakumu Kohosa (2000) .

NITE Chemical Risk Information Platform (NITE-CHRIP), National

Institute of Technology and Evaluation.

The information contained herein is based on several references and the present state of our knowledge. However the SDS does not always cover all information about the product, handle the product carefully. The information is intended to ordinary usage, in case of particular handlings, conduct appropriate safety measurements. The information herein is only provision of information, and it does not represent a guarantee the properties of the product. The Safety Data Sheet (SDS) is prepared based on JIS Z7253.